Table VIII. 4'-Fluoro-4-[4-arylcyclohexylamino]butyrophenone Hvdrochlorides

	x	NH(C	0 H <sub>2</sub> ) <sub>3</sub> C-	– F · HCl
Compd	Х	Yield, %	Mp, °C	Formula <sup><i>a</i></sup>
17a	p-F	43 <sup>b</sup>	193-197	C <sub>22</sub> H <sub>26</sub> ClF <sub>2</sub> NO <sup>C</sup>
17a	$p \cdot F^d$	49 <sup>e</sup>	190.5-195	$C_{22}H_{26}ClF_2NO^{f}$
17b	m-F	55	197-200	C <sub>22</sub> H <sub>26</sub> ClF <sub>2</sub> NO
17c	p-Cl	41	214-217	$C_{22}H_{26}Cl_2FNO^{C}$
17d	o-CH <sub>3</sub>	49	194-198	C <sub>23</sub> H <sub>29</sub> CIFNO <sup>g</sup>
17e	m-CH <sub>3</sub>	35 <sup>b</sup>	195-198	C <sub>23</sub> H <sub>29</sub> CIFNO
17f	p-CH <sub>3</sub>	41	227-230	$C_{23}H_{29}CIFNO^h$
17g	o-OCH <sub>3</sub>	45	223-235	C <sub>23</sub> H <sub>29</sub> CIFNO <sub>2</sub>
17ĥ	m-OCH <sub>3</sub>	58	183-185	C <sub>23</sub> H <sub>29</sub> CIFNO <sub>2</sub>
17i	p-OCH <sub>3</sub>	26 <sup>b</sup>	191-193	C <sub>23</sub> H <sub>29</sub> CIFNO <sub>2</sub>
17j	m-CF <sub>3</sub>	36 <sup>b</sup>	197-201	$C_{23}H_{26}CIF_4NO^{f}$
17k	p-CF <sub>3</sub>	26	206-210	$C_{23}H_{26}ClF_4NO^{f}$
171	<i>p</i> -F, <i>o</i> -CH <sub>3</sub>	52	215-218	C23H28ClF2NO·
				0.5H <sub>2</sub> O
17m	$p-NO_2$	46	231-232	$C_{22}H_{26}CIFN_2O$

<sup>a</sup>Analyses for C, H, N within 0.4% of theoretical values unless specified. <sup>b</sup>Recrystallized from MeOH-EtOAc. <sup>c</sup>Analysis for C, H only. <sup>d</sup>Cis compound. <sup>e</sup>Recrystallized from CH<sub>2</sub>Cl<sub>2</sub>-EtOAc. <sup>7</sup>Analysis for C, H, Cl. <sup>g</sup>Anal. Calcd: C, 70.84; H, 7.50. Found: C, 71.32; H, 8.09. <sup>h</sup>Satisfactory analysis could not be obtained; ir and nmr in agreement with structures.

MeOH was then removed in vacuo and the residue dissolved in 100 ml of hot H<sub>2</sub>O. The solution was then made strongly basic and extracted with Et<sub>2</sub>O. The organic layer was taken to dryness, the residue dissolved in a small amount of Et<sub>2</sub>O and this treated with 4.9 N HCl in Et<sub>2</sub>O. There was obtained 4.78  $\frac{2}{9}$  (99%) of hydrochloride. The analytical sample melted at 293-294°. Anal. (C<sub>12</sub>H<sub>17</sub>N<sub>2</sub>O) C, H, N.

trans-4-Arylcyclohexylpiperidine Tosylates (17) (Table VII). To a solution of 6.7 mmoles (1.64 g) of the amine hydrochloride in 30 ml of EtOH there was added 1.7 ml of 4.18 N NaOMe in MeOH. Following 1 hr stirring, there was added 1 ml (2.17 g) of 1,5-diiodopentane and 1.65 g of K<sub>2</sub>CO<sub>3</sub>. The mixture was stirred overnight at reflux and the solvent then removed in vacuo. The residue was taken up in H<sub>2</sub>O and Et<sub>2</sub>O (material insoluble in either phase was discarded). The organic layer was washed with H<sub>2</sub>O and brine and taken to dryness, in several cases this was recrystallized. The

residue was dissolved in ether and treated with 1 equiv of p-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H dissolved in ether. The precipitated tosyl salt was recrystallized from CH<sub>2</sub>Cl<sub>2</sub>-EtOAc.

General Procedure for Preparation of p-Fluorobutyrophenone (Table VIII). A suspension of 10 mmoles of the appropriate amine hydrochloride in 40 ml of DMF was treated with an equivalent (0.43 g) of 57% NaH in mineral oil. At the end of 1 hr there was added 1.70 g of KI, 2.82 g of  $K_2 CO_3$ , and 2.86 g of 4-chloro-pfluorobutyrophenone 2,2-dimethylpropylene ketal. The mixture was stirred overnight in an oil bath at 90°. The solvent was removed under oil pump vacuum and the residue dissolved in  $C_6H_6$  and  $H_2O$ . The organic layer was washed with H<sub>2</sub>O and brine and taken to drvness.

A mixture of the residue and 20 ml of 2.5 N HCl in 40 ml of MeOH was stirred at room temp for 2 hr. The bulk of the solvent was removed in vacuo and the solid collected on a filter. The cake was washed once with Et<sub>2</sub>O and recrystallized to constant mp from MeOH-2.5 N HCl.

trans.4'-Fluoro-4-[4-(p-fluorophenyl)cyclohexylmethylamino]butyrophenone hydrochloride (17n) was prepared in 60% yield by the above procedure. The product was recrystallized from CH<sub>2</sub>Cl<sub>2</sub>-EtOAc, mp 206-207.5°. Anal. (C23H28 CIF2 N) C, H, N.

cis-4'-Fluoro-4-[4-(p-fluorophenyl)cyclohexylmethylamino]butyrophenone hydrochloride (170) was prepared in 39% yield by the above procedure. The product was recrystallized from CH<sub>2</sub>Cl<sub>2</sub>-EtOAc to mp 191.5-193.5°. Anal. (C<sub>23</sub>H<sub>28</sub>ClF<sub>2</sub>N) C, H, N.

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# Central Nervous System Agents. 4.<sup>1</sup> Analogs of 3-Amino-2-phenylpropiophenone

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Thirty-one 1-aryl-2-phenyl-3-aminopropanones (I) have been tested in a battery of pharmacological screens designed to detect central nervous system activity. In general these Mannich bases seem to be stimulants. Most of these compounds and intermediates have not been previously reported.

Since the time of Mannich, scattered references have appeared to Mannich bases of the general formula I.<sup>2</sup> A con-



siderable number of this type of compound has been tested in animals in our laboratories. In general we have confirmed the reports by Huebner<sup>2g</sup> and Hofmann<sup>2h</sup> of CNS stimulant properties in this series, and we have extended the observations to many new compounds using a battery

of pharmacological tests. Table I lists the results of the most significant testing. That many of these compounds are stimulants is shown by the low dose at which stimulation was observed in intact mice as compared with the relatively high dose causing death or loss of righting reflex (generally >100-200 mg/kg). The potency is also indicated by the low dose which prevented nicotine-induced convulsions.

From these data it seems that the most potent stimulants have small substituents on the amine (H, Me, or cyclic with 3 or 4 carbons). While azetidine and pyrrolidine compounds are quite active (5, 6, 16, 17, 18, 19, and 26), compounds with the diethylamine group (4) or piperidine group (9) are much less active. It should be noted, however, that 9 was tested as a free base and thus may be more slowly absorbed



No.	Ar	Y	R	N~R'~``	нх	LD <sub>so</sub> . <sup><i>a</i></sup> mg/kg	Stimula- tion, <sup>a.b</sup> mg/kg	Traction, <sup>a</sup> Tr <sub>so</sub>	Chimney, <sup>a</sup> Ch <sub>50</sub>	Dish, <sup><i>a</i></sup> D <sub>50</sub>	Electro- shock, <sup>C</sup> PD <sub>50</sub>	Nicotine Antag, <sup>a</sup> AD <sub>50</sub>	Misc
1	C <sub>6</sub> H <sub>5</sub>	Н	Н	-NH <sub>2</sub>	HCl <sup>d</sup>	126	30	32	20	20	18	3.9	
23	C <sub>6</sub> H <sub>5</sub> C H	Н Н	н н	$-NHCH_3$ $-N(CH_3)$	C <sub>4</sub> H <sub>4</sub> O <sub>4</sub> " HC18	142 234	30	40 45	>25	12.5	16 20	12	$f_{f}$
4	C <sub>6</sub> H <sub>5</sub>	н	н	$-N(CH_2CH_3)_2$	HCl <sup>h</sup>	133	30	63	45	23	28	11	J
5	C <sub>s</sub> H <sub>s</sub>	Н	н	-NCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	C <sub>4</sub> H <sub>4</sub> O <sup>e</sup> <sub>4</sub>	100	10	89	50	45	>50	6.3	
6	C <sub>6</sub> H <sub>5</sub>	Н	Н	$-N(CH_2) CH_2$	HCl <sup>i</sup>	133	10	89	45	50	32	10	f
7	C <sub>6</sub> H <sub>5</sub>	Н	Н	-NCH <sub>2</sub> CH=CHCH <sub>2</sub>	HCl	200	10	71	>50	>50	71	20	
8	C <sub>6</sub> H <sub>5</sub>	Н	Н	$-N(CH_2)_3C(CH_3)_2$	HC1	142	10	56	45	40	>50	25	
9	C <sub>6</sub> H₅	Н	Н	$-N(CH_2)$ $CH_2$	Base <sup>j</sup>	562		>200	>200	>200	>200	57	
10	C <sub>6</sub> H <sub>5</sub>	Н	Н	-NCH2CH2SCH2CH2	Base	>10 <b>0</b> 0		>200	>200	112	>200	89	
11	C <sub>6</sub> H₅	Н	Н	-NCH2CH2N(CH3)CH2CH2	2HCl	178	30	100	63	>100	56	18	
1 <b>2</b>	C <sub>6</sub> H <sub>5</sub>	Н	Н	-NHCHCHCH2CH2NCH2CH2	Base	56	30	36	8	>50	>25	8	
13	C <sub>6</sub> H <sub>5</sub>	Н	Н	-NCH2CHCH2CH2CH2CH2CH2	Base	>1000	300	>200	178	126	>200	25	
14	C <sub>6</sub> H <sub>5</sub>	Н	Н	$-N(CH_2) \mathcal{L}(CH_2) \mathcal{L}H_2$	HCl	178		>200	126	100	200	18	
15	C <sub>6</sub> H <sub>5</sub>	н	Н	-NCH <sub>2</sub> CH <sub>2</sub> C $CH_2CH_2$ C $C_6H_5$	HCI	>1000	100	>200	89	>200	>200	79	
1 <b>6</b>	C₀ H₅	Н	CH₃	$-N(CH_2)_3CH_2$	HCI	142	10	63	50	>50	32	8	
17	C <sub>6</sub> H <sub>5</sub>	<i>m</i> -F	Н	$-N(CH_2)_{3}CH_2$	HCI	159	10	100	63	>100	56	18	
18	m-FC <sub>6</sub> H <sub>4</sub>	Н	Н	$-N(CH_2)_{3}CH_{2}$	HCI	142	10	63	>50	36	40	5.6	k
19	m-FC₅H₄	<i>m</i> -F	Н	$-N(CH_2)_3CH_2$	HCl	178	10	159	>100	36	72	18	k
20	p-ClC₅H₄	Н	н	-N(CH <sub>2</sub> ) <sub>3</sub> CH <sub>2</sub>	Base <sup>1</sup>	562	100	>200	>200	112	>200	63	
<b>2</b> 1	o-ClC ₀H₄	o-Cl	н	-N(CH <sub>2</sub> )₃CH <sub>2</sub>	HCI	178		159	100	18	178	79	
22	p-ClC₀H₄	p-Cl	н	$-N(CH_2)_3CH_2$	HCl	75	30	>200	50	72	>200	29	
23	p-(OH)C <sub>6</sub> H <sub>4</sub>	Н	Н	$-N(CH_2)_4CH_2$	Base <sup>m</sup>	>1000							
24 25	p-(OCH <sub>3</sub> )C <sub>6</sub> H <sub>4</sub> 3,4,5-(OCH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>2</sub>	<i>р</i> -(ОСН₃) Н	H H	-N(CH <sub>3</sub> ) <sub>2</sub> -N(CH <sub>2</sub> ) <sub>3</sub> CH <sub>2</sub>	HCI <sup>n</sup> HC1	75 142	30 30	72	50	45	>50	>50	f k
26		Н	н	$-N(CH_2)_3CH_2$	HCl	178	10	72	>50	>50	>50	23	0

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28	H	н	-N(CH <sub>3</sub> ) <sub>3</sub> -N(CH <sub>2</sub> ) <sub>3</sub> CH <sub>2</sub>	Base Base	>1000 750		>200 >200	>200 100	20 36	126 100
29 \ \\	Н	Н	-N(CH <sub>2</sub> ),CH <sub>2</sub>	Base	>1000	300 <i>P</i>	>200	>200	3.6	159
30 ( H	Н	Н	-NCH <sub>2</sub> CH <sub>2</sub> OCH <sub>2</sub> CH <sub>2</sub>	Base	>1000		> 200	>200	36	173
31)	Н	Н	-NCH <sub>2</sub> CH <sub>2</sub> N(CH <sub>3</sub> )CH <sub>2</sub> CH <sub>2</sub>	Base	1000	d	>200	126	28	50
<sup>a</sup> Carworth Farms m ip. Procedures for mea vulsions (TE) have dee	ale, albino mice suring acute tox n described prev	(CF-1) weig icity (LD 50) riously. <sup>3</sup> b7	hing 18-22 g were used for all studi ) and the effects of the compound of The lowest dose showing definite CP	les reported here. on overt behavior VS stimulation is	The test com , traction (Tr <sub>s</sub> , given. A blank	pds were diss ), chimney ( indicates no	olved or suspende (Ch <sub>50</sub> ), dish (D <sub>50</sub> ) stimulation was	ed in 0.25% aque , and antagonisn observed. <sup>c</sup> Proc	ous methylcellulose of nicotine-induce edures for the antag	soln and administered d tonic extensor con- onism of electro-

vulsions (TE) have deen described previously.<sup>3</sup>  $^{D}$ The lowest dose showing definite CNS stimulation is given. A blank indicates no stimulation was observed. <sup>c</sup>Procedures for the antagonism of electro-shock convulsions are described.<sup>4</sup> <sup>d</sup>See ref 5. <sup>e</sup>Maleate salt. <sup>f</sup>Anorexigenic activity was observed in dogs with a potency about  $^{1}$ /<sub>55</sub> that of amphetamine. <sup>g</sup>See ref 6. <sup>h</sup>See ref 2b. <sup>f</sup>See ref 7. <sup>j</sup>See ref 2a. <sup>f</sup>See ref 7. <sup>j</sup>See ref 7. <sup>j</sup>See ref 2a. <sup>f</sup>See ref 7. <sup>j</sup>See ref 2b. <sup>f</sup>See ref 7. <sup>j</sup>See ref 7. <sup>j</sup>See ref 2b. <sup>f</sup>See ref 7. <sup>j</sup>See ref 7. <sup>j</sup>See ref 2a. <sup>f</sup>See ref 7. <sup>j</sup>See ref 10. <sup>o</sup>In rat, ip LD<sub>50</sub> 141, stimulation at 12.5 mg/kg. <sup>p</sup>Depression was ob-served in some mice at 100 mg/kg. <sup>q</sup>Depression was observed at 300 mg/kg.

from the mouse peritoneum. An exception is compound 12, which contains a large amine moiety and is quite active. However, it is also quite toxic. Fluorine substituents on the benzene rings had little effect (17, 18, 19) but chlorine was detrimental (20, 21, and 22). The low CNS activity of the last five compounds in the table, containing the indole nucleus, is surprising in view of their relationship to CNS active indoles such as  $\alpha$ -ethyltryptamine (etryptamine) which shows CNS stimulation in mice at 30 mg/kg and antagonizes nicotine-induced tonic extensor convulsions at 5.6 mg/kg. A few of the compounds (9, 14, and 19) showed some promise as anorexigenics having activity on the order of 0.1 that of amphetamine.

These compounds were synthesized from the corresponding aryl benzyl ketone either directly by the Mannich reaction<sup>2a</sup> (methods A or C) or indirectly by the addition of an amine to 2-phenylacrylophenone<sup>2e,7</sup> (method B) which was in turn prepared *via* a Mannich base. The latter method usually gave better results. These methods are illustrated in the Experimental Section and the new compounds are given in Table II or in the Experimental Section.

### Experimental Section †

Method B. 3-(1-Azetidinyl)-2-phenylpropiophenone (5). To 13.2 g (0.15 mole) of 2-phenylacrylophenone<sup>7</sup> was added 6.8 ml (0.1 mole) of azetidine giving a soln which soon crystallized. After standing for 7 hr, the solid was recrystallized from 75 ml of hexane yielding 20.3 g (77%) of white crystals, mp  $80.5-82^{\circ}$ .

*d*-2-Phenyl-3-(1-pyrrolidinyl)propiophenone Hydrochloride (6d). The *d*-10-camphorsulfonate salt of 2-phenyl-3-(1-pyrrolidinyl)propiophenone was prepd in EtOAc from 168 g (0.6 mole) of *dl* base<sup>2</sup> and 139.5 g (0.6 mole) of *d*-10-camphorsulfonic acid. This was recrystallized 11 times from EtOAc giving 2.6 g of white cryst, mp 145-146.5°,  $[\alpha]^{25}D$ +61.6 ± 2° ( $\alpha_0^{25}$ +1.35 ± 0.03° (*c* 0.0219 g in 2 ml of MeOH). Anal. C, H, N, S.

A soln of 2.1 g of this salt was converted to the hydrochloride on a column of 100 ml of Amberlite XE-208 resin in the chloride form. The product was eluted with 90% EtOH giving 1.3 g of crystalline solid,  $[\alpha]^{25}D+65^{\circ}$ . This was recrystallized three times from EtCOMe yielding 0.16 g of white crystals, mp 157-159°,  $[\alpha]^{25}D+171 \pm 3^{\circ} (\alpha_0^{25}+3.45 \pm 0.05^{\circ}, c \ 0.0202 \ g \ in 2 \ ml of MeOH)$ . The rotation remained unchanged after the last recrystallization.

2-Phenyl-3-(3-quinuclidinylamino)propiophenone (12). To a soln of 25 g (0.126 mole) of 3-aminoquinuclidine dihydrochloride and 25.4 g (0.252 mole) of  $Et_3N$  in 100 ml of MeOH was added 26.2 g (0.126 mole) of 2-phenylacrylophenone. After standing for 5 days, the soln was concd to dryness and the residue was dissolved in cold dil HCl. The acid soln was washed with  $Et_2O$  and basified with cold NaOH. The resulting oil was extd with  $Et_2O$ , washed (water and sat NaCl soln), and dried  $(Na_2SO_4)$ . Filtration and removal of the ether gave 36 g of a syrup which crystallized on standing, mp 90-95°. This was recrystallized from EtOAc yielding 19 g (45%) of white crystals, mp 98.5-100°.

2-Methyl-2-phenyl-3-(1-pyrrolidinyl)propiophenone Hydrochloride (16). To a cold soln of 7.8 ml (0.1 mole) of pyrrolidine in 100 ml of 95% EtOH was slowly added 7.2 ml (0.09 mole) of 37% aqueous CH<sub>2</sub>O, followed by 15.5 g (0.373 mole) of 2-phenylpropiophenone. The soln was heated under reflux for 22 hr, dild with water, and extd with Et<sub>2</sub>O. The Et<sub>2</sub>O soln was washed (H<sub>2</sub>O and satd NaCl soln) and dried (K<sub>2</sub>CO<sub>3</sub>). After filtration, the soln was acidified with ethanolic HCl. The resulting oil was separated and crystallized on shaking with Et<sub>2</sub>O. Two recrystns from *i*-PrOH yielded 2.94 g (12.2%) of white crystals, mp 194-196.5°. 2-Phenylpropiophenone.‡ A soln of 20.8 g (0.1 mole) of 2-

2-1 neny proproprienone. + A som of 20.8 g (0.1 mole) of 2-

 $\ddagger$  This was first prepared by Meyer and Oelkers<sup>11</sup> and has since been made by numerous methods. The present new synthesis seems to be one of the most convenient.

<sup>†</sup>Melting points were taken in capillary tubes with a partial immersion thermometer. Calibration of the apparatus against standard compounds showed no need for correction. Ir (Nujol mull) and usually nmr spectra (in DMSO- $d_6$  or CDCl<sub>3</sub>) were obtained on pure compounds and were in accordance with the proposed structure. Where analyses are indicated only by symbols of the elements of functions, analytical results obtained for these elements or functions were within ±0.4% of theoretical.

#### Table II. Chemistry



No. <sup>a</sup>	Method of prepn	Yield, % <sup>b</sup>	Мр, °С <sup>с</sup>	Crystallization solvent	Formula	Anal. <sup>c</sup>
2	Bd	22	134.5-135.5	<i>i</i> -PrOH	C. H. NO.	C.H.N
- 5b	B	77	80.5-82	Hexane	C.,H.,NO	C. H. N
5	е	96	135-136	i-PrOH	C.H.NO.	C. H. N
<b>6</b> d <sup><i>f</i></sup>	g		157-159	EtCOMe	C. H. CINO	C. H
7b	B	96	99-101	i-PrOH	C. H. NO	C. H. N
7	h	93	151-154.5	<i>i</i> -PrOH	C. H. CINO	C. H. Cl. N
8	$\mathbf{B}^{i}$	82	153-156	i-PrOH-Et <sub>2</sub> O	C <sub>a</sub> H <sub>a</sub> ClNO	C. H. Cl. N
10	В	89	106-107	<i>i</i> -PrOH	C. H. NOS	C. H. N. S
11	Bį	87	203-207 dec	EtOH	CarHaclaNa O	C, H, Cl, N
1 <b>2</b>	$\mathbf{B}^{\mathbf{g}}$	45	98,5-100	EtOAc	CarHanNaO	C. H. N
13	В	84	113-115	EtOH	CarHanNO	C, H, N
14b	В	60	71-72.5	<i>i</i> -PrOH	CarHanNO	C. H. N
14	h	96	169-170	<i>i</i> -PrOH	C, H <sub>2</sub> CINO	C, H, Cl, N
1 <b>5</b> b	В	81	142.5-144.5	EtOH	C,H,NO,	C, H, N
15	h	75	201-202.5	MeOH-Et <sub>2</sub> O	C,H,CINO,	C, H, CI, N
1 <b>6</b>	Ag	12.5	194.5-196.5	<i>i</i> -PrOH	C <sub>20</sub> H <sub>24</sub> CINO	C, H, CI, N
17b	Ag	50	102.5-104	95% EtOH	C <sub>10</sub> H <sub>20</sub> FNO	C, H, F, N
17	k	77	166.5-168	EtOH	C1.H,CIFNO	C, H, CI, F, N
1 <b>8</b> b	Α	42	85-86.5	95% EtOH	C <sub>1</sub> H <sub>20</sub> FNO	C, H, F, N
18	k	82	159-160	<i>i</i> -PrOH	C, H, CIFNO	C, H, Cl, F, N
19b	Α	33	85-86	Hexane	C <sub>1</sub> H <sub>1</sub> F <sub>2</sub> NO	C, H, F, N
19	k	58	153-154	<i>i</i> -PrOH	C <sub>19</sub> H <sub>20</sub> CIF <sub>2</sub> NO	C, H, Cl, F, N
<b>2</b> 1	$\mathbf{B}^{g}$	40	135-136	EtCOMe	C <sub>19</sub> H <sub>20</sub> Cl <sub>3</sub> NO	C, H, Cl, N
22	$B^{l}$	96	161-162	EtCOMe	C <sub>19</sub> H <sub>20</sub> Cl <sub>3</sub> NO	C, H, Cl, N
<b>25</b> b	Α	44	112.5-114	<i>i</i> -PrOH	$C_{22}H_{27}NO_4$	C, H, N
25	k	97	168-169.5 dec	95% EtOH	C <sub>22</sub> H <sub>28</sub> CINO <sub>4</sub>	C, H, N, Cl
<b>26</b> b	$A^m$	100	119.5-121	MeOH	C <sub>17</sub> H <sub>19</sub> NOS	C, H, N, S
26	k	90	168-170	<i>i</i> -PrOH	C <sub>17</sub> H <sub>20</sub> CINOS	C, H, Cl, N, S
<b>2</b> 7	$C^n$	39	186.5-187.5	EtOAc	$C_{19}H_{20}N_2O$	C, H, N
28	C <sup>o</sup>	49 <sup>0</sup>	141.5-144.5	EtOAc	$C_{21}H_{22}N_{2}O$	C, H, N
<b>2</b> 9	Cg	54	172.5-173.5	EtOAc-hexane	$C_{22}H_{24}N_{2}O$	C, H, N
30	$C^p$	6.1	167-169	EtOAc	$C_{21}H_{22}N_{2}O_{2}$	C, H, N
31	Ba	47	183.5-184.5	EtOAc-hexane	C <sub>22</sub> H <sub>25</sub> N <sub>3</sub> O	C, H, N

<sup>a</sup>Numbers correspond to those in Table I. A "b" after the number indicates a free base corresponding to the salt of Table I. <sup>b</sup>Yields of salts are calcd from the Mannich bases if the bases were purified. Yields of free bases (or salts if bases were not purified) are calcd from the corresponding desoxybenzoin (methods A or C) or from the corresponding 2-phenylacrylophenone (method B). Unless otherwise indicated, yields are reported for material melting not less than 2° below the highest mp obtd. <sup>c</sup>See footnote  $\dagger$ . <sup>d</sup>The free base failed to crystallize. A slight excess of maleic acid was added to an ethereal soln of the crude base and the resulting maleate was recrystd from *i*-PrOH. The hydrochloride has been reported, ref 2e. <sup>e</sup>A warm soln of 5b and a slight excess of maleic acid in *i*-PrOH crystallized on cooling giving the maleate salt. <sup>f</sup> 6d is the dextro (+) rotating optical isomer corresponding to 6 of Table I. <sup>g</sup>Preparation described in the Experimental Section. <sup>h</sup>The free base was converted to the hydrochloride in *i*-PrOH with a slight excess of ethanolic HCl. <sup>i</sup>The free base failed to crystallize even after standing for 6 weeks. It was converted to the hydrochloride in *i*-PrOH with ethanolic HCl and diluted with Et<sub>2</sub>O. <sup>j</sup>The free base was converted to the dihydrochloride in *E*OH. <sup>k</sup>Prepared from the free base in abs Et<sub>2</sub>O with ethanolic HCl. <sup>l</sup>Prepared as described in the Experimental Section for 21. <sup>m</sup>The Mannich base cryst from the reaction mixt and was collected, washed (*i*-PrOH), and dried. <sup>n</sup>Dimethylamine hydrochloride and sodium acetate were used in place of the free amine. About 13% of 1-(indol-3-yl)-2-phenyl-1-propenone was isolated. <sup>o</sup>About 35% of 1-(indol-3-yl)-2-phenyl-1-propenone and 49% of crude 27 were isolated. <sup>p</sup>A yield of 51.2% of 1-(indol-3-yl)-2-phenyl-1-propenone was isolated.

phenylacrylophenone<sup>7</sup> in 100 ml of MeOH was hydrogenated with 0.2 g of PtO<sub>2</sub> at 3.5 kg/cm<sup>2</sup> and room temp. After filtration and evaporation, the product was crystallized from 50 ml of pentane yielding 15.8 g (75%) of white crystals, mp 48-50°. 2-(m-Fluorophenyl)acetophenone.<sup>§</sup> m-Fluorobenzylmagnesium

2-(*m*-Fluorophenyl)acetophenone. § *m*-Fluorobenzylmagnesium chloride was prepd from 18.2 g (0.75 g-atom) of Mg, 108 g (0.75 mole) of *m*-fluorobenzyl chloride, and 375 ml of abs  $Et_2O$ . To this was slowly added 51.5 g (0.5 mole) of benzonitrile in 375 ml of abs  $Et_2O$ . After refluxing 2.5 hr and standing overnight, the mixt was poured onto ice and 120 ml of concd HCl. After standing for 1 hr, the product was extd with ether and  $CH_2Cl_2$ , washed (H<sub>2</sub>O and satd NaCl), and dried (Na<sub>2</sub>SO<sub>4</sub>). Filtration and evaporation gave 131.3 g of crude product, mp 45-48°. Recrystallization from 160 ml of methylcyclohexane yielded 97.3 g (92%) of crystals, mp 47.5-49.5°

Method A. 2-(*m*-Fluorophenyl-3-(pyrrolidinyl)propiophenone (17b). A soln of 21.4 g (0.1 mole) of 2-(*m*-fluorophenyl)aceto-phenone, 12 ml of 37% aqueous CH<sub>2</sub>O, and 9.5 ml (0.122 mole) of pyrrolidine in 100 ml of MeOH was allowed to stand for 3 days at

room temp and evapd to dryness *in vacuo*. The residue was dissolved in dil HCl, washed (ether), and basified with dil NaOH. The resulting oil crystallized and was collected, washed (H<sub>2</sub>O), dried, and recrystallized from 90 ml of 95% EtOH yielding 14.7 g (50%) of crystals, mp 102.5-104°.

3'-Fluoro-2-phenylacetophenone. This was prepd as the above 2-(*m*-fluorophenyl)acetophenone from 19.4 g (0.8 g-atom) of Mg, 100.8 g (0.8 mole) of benzyl chloride, 50 g (0.415 mole) of *m*-fluorobenzonitrile, and 750 ml of abs Et<sub>2</sub>O. The product was recrystallized from hexane yielding 77.2 g (87%) of white crystals, mp 63-65°. Anal. C, H, F.

3'-Fluoro-2-(*m*-fluorophenyl)acetophenone. This was prepd as the above 2-(*m*-fluorophenyl)acetophenone from 19.4 g (0.8 g-atom) of Mg, 115 g (0.8 mole) of *m*-fluorobenzyl chloride, 50 g (0.415 mole) of *m*-fluorobenzonitrile, and 750 ml of abs  $Et_2O$ . The product was recrystallized from methylcyclohexane yielding 70 g (70%) of crystals, mp 49-50.5°. Anal. C, H; F, calcd 16.36, found 17.01.

**2-Chloro-2-(o-chlorophenyl)acrylophenone.** A soln of 53 g (0.2 mole) of 2'-chloro-2-(o-chlorophenyl)acetophenone, <sup>13</sup> 48 ml (0.6 mole) of 37% CH<sub>2</sub>O, and 1 ml of pyrrolidine in 160 ml of

<sup>§</sup> This has been reported by Fischer, et al, <sup>12</sup> mp 47°, prepd by a different method. No yield of pure material was given.

MeOH was refluxed for 4 hr and allowed to stand for 3 days. Water was added, and the mixt was extd with  $Et_2O$ . The  $Et_2O$  soln was washed (dil HCl, dil NaHCO<sub>3</sub>, H<sub>2</sub>O, and satd NaCl) and dried (Na<sub>2</sub>SO<sub>4</sub>). After filtration and evaporation of the solvent, the product was distd giving 36.8 g of solid, bp 152° (0.1 mm). A sample was recrystallized from *i*-PrOH giving white crystals, mp 81-83°. Anal. C, H, Cl.

2'-Chloro-2-(o-chlorophenyl)-3-(1-pyrrolidinyl)propiophenone Hydrochloride (21). A mixt of 35 g (0.126 mole) of 2'-chloro-2-(ochlorophenyl)acrylophenone and 12 ml (0.15 mole) of pyrrolidine was warmed to effect soln and allowed to stand for 4 days. The resulting oil was dissolved in Et<sub>2</sub>O, washed (H<sub>2</sub>O, satd NaCl), and dried (Na<sub>2</sub>SO<sub>4</sub>). After filtration, the soln was acidified with ethanolic HCl, and the resulting solid was crystallized from EtCOMe yielding 19.6 g of white crystals, mp 135-136°.

4<sup>3</sup>Chloro-2-(*p*-chlorophenyl)acrylophenone. This was prepd as described above for the ortho isomer from 10 g (0.0378 mole) of 4-chloro-2-*p*-chlorophenyl)acetophenone,<sup>13</sup> 8.8 ml (0.11 mole) of 37% CH<sub>2</sub>O, and 0.19 ml of piperidine in 35 ml of MeOH. The product was not distilled but was recrystallized from hexane yielding 7.2 g (69%) of white crystals, mp 80-83°. Anal. C, H, Cl

Benzyl Indol-3-yl Ketone. A soln of 287.6 g of N,N-dimethylphenylacetamide in 300 ml of PhH was cooled under N<sub>2</sub> to 10° and 108 ml of POCl<sub>3</sub> was added dropwise with stirring. After warming to 20°, 102.9 g (0.878 mole) of indole was slowly added keeping the temp below 48° by cooling. The mixt was then refluxed for 2 hr, cooled, and poured into 6 l. of H<sub>2</sub>O. A soln of 307.5 g of NaOH in 914 ml of H<sub>2</sub>O was added, and the mixt was stirred for 1 hr. The product was extd with Et<sub>2</sub>O, washed (H<sub>2</sub>O, satd NaCl), and evapd *in* vacuo. The residue was mixed with 200 g of NaOAc in 200 ml of H<sub>2</sub>O and 3 l. of MeOH, refluxed for 3 hr, and cooled to near 0°. The solid was collected, washed (MeOH), and dried yielding 122.6 g of product, mp 206.5-209°. An addnl 45.5 g (total 81.5%) was obtained by concn of the filtrate. A sample was recrystd from EtOAc, mp 208-209°. Anal. C, H, N.

Method C. 1-(Indoi-3-yl)-2-phenyl-1-propenone and 1-(Indoi-3-yl)-2-phenyl-3-(1-piperidinyl)-1-propanone (29). A mixt (prepared under N<sub>2</sub> with cooling) of 50 ml of AcOH, 4.9 ml (0.05 mole) of piperidine, 11.76 g (0.05 mole) of benzyl indoi-3-yl ketone, and 2 g (0.0666 mole) of paraformaldehyde was stirred at 100-110° for 3-5 hr, cooled, and concd *in vacuo* to half its vol. This was poured into ice water and the solid was collected, washed (H<sub>2</sub>O), and dried giving 4.68 g of 1-(indoi-3-yl)-2-phenyl-1-propenone, mp 175-181.5°. Recrystallization from EtOAc-hexane gave 1.17 g of crystals, mp 195-196.5°. Anal. C, H, N.

The aqueous filtrate was basified with NaOH giving free base which was collected, dried, and recrystallized from EtOAc yielding 8.99 g (54%) of 29, mp 171.5-172.5°. Recrystallization from EtOAc-hexane raised the mp to  $172.5-173.5^{\circ}$ .

1-(Indol-3-yl)-3-(4-methyl-1-piperazinyl)-2-phenyl-1-propanone (31). A mixt of 6.33 g (0.0266 mole) of 1-(indol-3-yl)-2-phenyl-1propenone and 20 ml of 1-methylpiperazine was heated under N<sub>2</sub> at 100° for 3 hr and poured into ice water. The resulting solid was dissolved in CHCl<sub>3</sub> and extd with dil HCl. The acid soln was basified with NaOH giving solid which was collected, washed (H<sub>2</sub>O), dried, and recrystallized first from EtCOMe-cyclohexane and then from EtOAc-hexane yielding 4.3 g of 31, mp 183.5-184.5°.

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## Potential Bioreductive Alkylating Agents. 1. Benzoquinone Derivatives

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A series of benzoquinone derivatives with one or two side chains potentially capable of alkylation after bioreduction was synthesized. These compounds showed growth-inhibitory activity against adenocarcinoma 755 ascites cells and greatly prolonged the life-span of such tumor-bearing mice. Compounds of this series were also found to be potent inhibitors of the synthesis of both DNA and RNA in these neoplastic cells.

Mitomycin C, an antineoplastic agent active against tumors of both animals and man, has been shown to be a strong inhibitor of the synthesis of the nucleic acids.<sup>1</sup> Iyer and Szybalsky<sup>2</sup> presented evidence to indicate that mitomycins act as bifunctional alkylating agents which add across both strands of the DNA double helix to cause cross-linking. Schwartz, *et al.*,<sup>3</sup> demonstrated that the reduction of the benzoquinone ring of the mitomycins to dihydrobenzoquinone was an essential step for biological activity and Iyer and Szybalsky<sup>4</sup> showed that a NADPH (reduced form of nicotinamide-adenine dinucleotide phosphate) dependent quinone reductase system was involved in the reductive activation step. Recently, Kinoshita and coworkers<sup>5</sup> reported a positive correlation between both antineoplastic and antimicrobial activity of a series of mitomycin derivatives and their reduction potentials. These investigators<sup>5</sup> also provided evidence that the carbamyl group and the aziridine ring of the mitomycins were not essential for biological activity. The essential portions of the mitomycins were proposed to be the structures shown in formulas I and II (Scheme I). It is possible that charge delocalization of the dihydroquinone hydroxyl